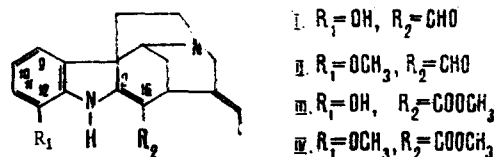


STRUCTURE OF VINERVININE, VINERVINE,  
VINCANIDINE, AND VINCANICINE

M. R. Yagudaev, V. M. Malikov,  
and S. Yu. Yunusov

UDC 547.944.1

Previously, for the alkaloids vincanidine [1-3], vincanicine [4], vinervine [3, 5], and vinervinine [6], mainly on the basis of their IR spectroscopic characteristics [1], structures I, II, III, and IV, respectively, with the substituent  $R_1$  in the benzene ring and  $C_{11}$  have been proposed.



To determine the positions of the substituents OH and  $OCH_3$  in the aromatic nucleus of (I-IV) we have studied the NMR spectra of (I), (II), and (IV) and some derivatives of (IV) (JNM-4H-100/100 MHz spectrometer,  $CDCl_3$ , HMDS-10). In the NMR spectra of (I), (II), and (IV) the signals from the three protons of the aromatic ring give a multiplet (3H) in the range  $\tau$  3.2-3.5 ppm. In the spectrum of 2,16-dihydrovinervinine (V) (Fig. 1a), the signals of these protons are also observed in the form of a multiplet. Further, in the NMR spectra of the 11-methoxy substituted  $\alpha$ -methyleneindoline alkaloids 16-methoxyvincadifformine (ervinceine) and its 2,3-dihydro derivative, as in the spectrum of 16-methoxytabersonine (ervamicine) [7], the signals of the aromatic protons formed a three-spin system of the ABC type the  $H_9$  signal in which is located in a relatively weak field as compared with the  $H_{10}$  and  $H_{12}$  signals.

Moreover, in indoline alkaloids with a substituent in position 11 of the aromatic ring the signal of the  $H_{12}$  proton, which has spin-spin coupling with  $H_{10}$  (meta constant), shifts by 1.0 ppm and more on the acetylation of the N-H [8, 9]. In the NMR spectrum of N-acetyldihydrovinervinine (VI) (Fig. 1b) the three aromatic protons reveal a pattern of the ABC type, and the signal of one of them appears in the form of a quartet with its center at 2.93 ppm and SSCCs of  $J_1 = 8.0$  and  $J_2 = 8.5$  Hz. The values of the SSCCs, corresponding to ortho interaction of aromatic protons, show that the quartet with its center at 2.93 ppm must be assigned to the signal of a proton in ortho positions to which there are two other protons. Consequently, the  $OCH_3$  group in (VI) and also the substituents  $OCH_3$  and OH in the initial alkaloids (I-IV) cannot be located at  $C_{11}$  of the aromatic ring, as was considered previously. Substitution at  $C_9$  for the OH and  $OCH_3$  groups is excluded, since the expected signal of the  $H_{12}$  proton (quartet with  $J_{ortho} \approx 8.0$  and  $J_{meta} \approx 2.0$ ) in the spectrum of (VI) is in practically the same position as the CS of  $H_{12}$  in the initial spectrum of (V) (see Fig. 1a, b).

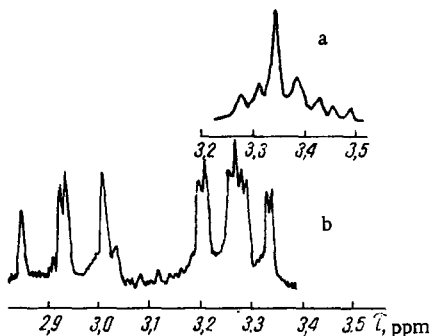


Fig. 1. NMR spectrum of 2,16-dihydrovinervinine (V) (a) and of N-acetyldihydrovinervinine (VI) (b).

Thus, the results of a detailed consideration of the CSs and SSCCs of the signals of the aromatic protons in the NMR spectra of vinervinine and its dihydro and N-acetyldihydro derivatives unambiguously show that the methoxy group in them is located at  $C_{12}$ . These results show that in vinervine, vincanidine, and vincanicine, likewise, the OH and  $OCH_3$  groups are located at  $C_{12}$  of the aromatic ring.

Order of the Red Banner of Labor Institute of the Chemistry of Plant Substances, Academy of Sciences of the Uzbek SSR. Translated from *Khimiya Prirodnikh Soedinenii*, No. 2, pp. 260-261, March-April, 1974. Original article submitted October 21, 1973.

© 1975 Plenum Publishing Corporation, 227 West 17th Street, New York, N.Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$15.00.

#### LITERATURE CITED

1. S. Yu. Yunusov and P. Kh. Yuldashev, Dokl. Akad. Nauk UzSSR, No. 12, 24 (1952); Zh. Obshch. Khim., 27, 2015 (1957).
2. Kh. Ubaev, P. Kh. Yuldashev, and S. Yu. Yunusov, Dokl. Akad. Nauk, UzSSR, No. 10, 34 (1964); Izv. Akad. Nauk SSSR, Ser. Khim., 1992 (1965).
3. P. Kh. Yuldashev, Kh. Ubaev, M. A. Kuchenkova, and S. Yu. Yunusov, Khim. Prirodn. Soedin., 34 (1965).
4. D. A. Rakhimov, V. M. Malikov, and S. Yu. Yunusov, Khim. Prirodn. Soedin., 461 (1969).
5. M. A. Kuchenkova, P. Kh. Yuldashev, and S. Yu. Yunusov, Dokl. Akad. Nauk UzSSR, No. 11, 42 (1964); Izv. Akad. Nauk SSSR, Ser. Khim., 12, 2152 (1965).
6. N. Abdurakhimova, P. Kh. Yuldashev, and S. Yu. Yunusov, Khim. Prirodn. Soedin., 310 (1967); Dokl. Akad. Nauk SSSR, 173, 87 (1967).
7. D. A. Rakhimov, V. M. Malikov, M. R. Yagudaev, and S. Yu. Yunusov, Khim. Prirodn. Soedin., 226 (1970).
8. S. McLean et al., Can. J. Chem., 38, 1547, 2278 (1960).
9. K. Nagarajan et al., Tetrahedron, 23, 1683 (1967).